SAFETY DATA SHEET

Version 4.14 Revision Date 04/13/2018 Print Date 10/04/2018

1. PRODUCT AND COMPANY IDENTIFICATION

1.1 Product identifiers

Product name : Lead(II) fluoride

Product Number : 236152
Brand : Aldrich
Index-No. : 082-001-00-6

CAS-No. : 7783-46-2

1.2 Relevant identified uses of the substance or mixture and uses advised against

Identified uses : Laboratory chemicals, Synthesis of substances

1.3 Details of the supplier of the safety data sheet

Company : Sigma-Aldrich

3050 Spruce Street

SAINT LOUIS MO 63103

USA

Telephone : +1 800-325-5832 Fax : +1 800-325-5052

1.4 Emergency telephone number

Emergency Phone # : +1-703-527-3887 (CHEMTREC)

2. HAZARDS IDENTIFICATION

2.1 Classification of the substance or mixture

GHS Classification in accordance with 29 CFR 1910 (OSHA HCS)

Acute toxicity, Oral (Category 2), H300 Acute toxicity, Inhalation (Category 2), H330

Carcinogenicity (Category 1B), H350 Reproductive toxicity (Category 1A), H360

Specific target organ toxicity - repeated exposure (Category 2), H373

Acute aquatic toxicity (Category 1), H400 Chronic aquatic toxicity (Category 1), H410

For the full text of the H-Statements mentioned in this Section, see Section 16.

2.2 GHS Label elements, including precautionary statements

Pictogram



Signal word Danger

Hazard statement(s)

H300 + H330 Fatal if swallowed or if inhaled.

H350 May cause cancer.

H360 May damage fertility or the unborn child.

H373 May cause damage to organs through prolonged or repeated exposure.

H410 Very toxic to aquatic life with long lasting effects.

Precautionary statement(s)

P201 Obtain special instructions before use.

P202 Do not handle until all safety precautions have been read and understood. P260 Do not breathe dust/ fume/ gas/ mist/ vapours/ spray. P264 Wash skin thoroughly after handling. Do not eat, drink or smoke when using this product. P270 Use only outdoors or in a well-ventilated area. P271 Avoid release to the environment. P273 Wear protective gloves/ protective clothing/ eye protection/ face P280 protection. Wear respiratory protection. P284 P301 + P310 + P330 IF SWALLOWED: Immediately call a POISON CENTER/doctor. Rinse mouth. IF INHALED: Remove person to fresh air and keep comfortable for P304 + P340 + P310 breathing. Immediately call a POISON CENTER/doctor. IF exposed or concerned: Get medical advice/ attention. P308 + P313 P391 Collect spillage. P403 + P233 Store in a well-ventilated place. Keep container tightly closed. P405 Store locked up.

Dispose of contents/ container to an approved waste disposal plant.

Hazards not otherwise classified (HNOC) or not covered by GHS

Strong hydrogen fluoride-releaser

3. COMPOSITION/INFORMATION ON INGREDIENTS

3.1 Substances

2.3

P501

Formula : F₂Pb

Molecular weight : 245.20 g/mol CAS-No. : 7783-46-2 EC-No. : 231-998-8 Index-No. : 082-001-00-6

Hazardous components

Component	Classification	Concentration	
Lead difluoride			
	Acute Tox. 2; Carc. 1B; Repr. 1A; STOT RE 2; Aquatic Acute 1; Aquatic Chronic 1; H300 + H330, H350, H360, H373, H410	90 - 100 %	

For the full text of the H-Statements mentioned in this Section, see Section 16.

4. FIRST AID MEASURES

4.1 Description of first aid measures

General advice

Hydrofluoric (HF) acid burns require immediate and specialized first aid and medical treatment. Symptoms may be delayed up to 24 hours depending on the concentration of HF. After decontamination with water, further damage can occur due to penetration/absorption of the fluoride ion. Treatment should be directed toward binding the fluoride ion as well as the effects of exposure. Skin exposures can be treated with a 2.5% calcium gluconate gel repeated until burning ceases. More serious skin exposures may require subcutaneous calcium gluconate except for digital areas unless the physician is experienced in this technique, due to the potential for tissue injury from increased pressure. Absorption can readily occur through the subungual areas and should be considered when undergoing decontamination. Prevention of absorption of the fluoride ion in cases of ingestion can be obtained by giving milk, chewable calcium carbonate tablets or Milk of Magnesia to conscious victims. Conditions such as hypocalcemia, hypomagnesemia and cardiac arrhythmias should be monitored for, since they can occur after exposure. Consult a physician. Show this safety data sheet to the doctor in attendance. Move out of dangerous area.

Aldrich - 236152 Page 2 of 9

If inhaled

If breathed in, move person into fresh air. If not breathing, give artificial respiration. Consult a physician.

In case of skin contact

First treatment with calcium gluconate paste. Wash off with soap and plenty of water. Take victim immediately to hospital. Consult a physician.

In case of eye contact

Flush eyes with water as a precaution.

If swallowed

Never give anything by mouth to an unconscious person. Rinse mouth with water. Consult a physician.

4.2 Most important symptoms and effects, both acute and delayed

The most important known symptoms and effects are described in the labelling (see section 2.2) and/or in section 11

4.3 Indication of any immediate medical attention and special treatment needed

No data available

5. FIREFIGHTING MEASURES

5.1 Extinguishing media

Suitable extinguishing media

Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.

5.2 Special hazards arising from the substance or mixture

No data available

5.3 Advice for firefighters

Wear self-contained breathing apparatus for firefighting if necessary.

5.4 Further information

No data available

6. ACCIDENTAL RELEASE MEASURES

6.1 Personal precautions, protective equipment and emergency procedures

Wear respiratory protection. Avoid dust formation. Avoid breathing vapours, mist or gas. Ensure adequate ventilation. Evacuate personnel to safe areas. Avoid breathing dust.

For personal protection see section 8.

6.2 Environmental precautions

Prevent further leakage or spillage if safe to do so. Do not let product enter drains. Discharge into the environment must be avoided.

6.3 Methods and materials for containment and cleaning up

Pick up and arrange disposal without creating dust. Sweep up and shovel. Keep in suitable, closed containers for disposal.

6.4 Reference to other sections

For disposal see section 13.

7. HANDLING AND STORAGE

7.1 Precautions for safe handling

Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Further processing of solid materials may result in the formation of combustible dusts. The potential for combustible dust formation should be taken into consideration before additional processing occurs.

Provide appropriate exhaust ventilation at places where dust is formed.

For precautions see section 2.2.

7.2 Conditions for safe storage, including any incompatibilities

Keep container tightly closed in a dry and well-ventilated place.

Do not store in glass Keep in a dry place.

Storage class (TRGS 510): 6.1D: Non-combustible, acute toxic Cat.3 / toxic hazardous materials or hazardous materials causing chronic effects

Aldrich - 236152 Page 3 of 9

7.3 Specific end use(s)

Apart from the uses mentioned in section 1.2 no other specific uses are stipulated

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

8.1 Control parameters

Components with workplace control parameters

Component	CAS-No.	Value	Control	Basis				
Lead difluoride	7783-46-2	TWA	parameters 2.5 mg/m3	LISA Cogunational Exposure Limita				
Lead dilluoride	1103-40-2	IVVA	2.5 mg/m3	USA. Occupational Exposure Limits (OSHA) - Table Z-1 Limits for Air				
				Contaminants				
	Domorko	CAS number varies with compound						
	Remarks							
		IVVA	2.5 mg/m3	USA. ACGIH Threshold Limit Values (TLV)				
		Bone damag	ge					
		Fluorosis						
		Substances for which there is a Biological Exposure Index or Indices						
		(see BEI® section) Not classifiable as a human carcinogen						
			arcinogen					
		varies						
		TWA	0.05 mg/m3	USA. ACGIH Threshold Limit Values (TLV)				
		Central Nervous System impairment						
		Hematologic effects						
			Peripheral Nervous System impairment					
		Substances for which there is a Biological Exposure Index or Indices						
		(see BEI® section)						
		Confirmed animal carcinogen with unknown relevance to humans						
		varies						
		PEL	0.05 mg/m3	OSHA Specifically Regulated				
				Chemicals/Carcinogens				
			1910.1025					
			If an employee is exposed to lead for more than 8 hours in any work					
				limit, as a time weighted average				
		(TWA) for that day, shall be reduced according to the following						
		formula: Maximum permissible limit (in μg/m3)=400÷hours worked						
		in the day						
		This section applies to all occupational exposure to lead, except as						
		provided in paragraph (a)(2). It does not apply to the construction						
			industry or to agricultural operations covered by 29 CFR part 1928.					
		OSHA specifically regulated carcinogen						
		TWA	0.05 mg/m3	USA. NIOSH Recommended				
		Coo Arrier : 1	iv C	Exposure Limits				
		See Appendix C						
		PEL	2.5 mg/m3	California permissible exposure				
				limits for chemical contaminants				
		DEL	0.05 1:0	(Title 8, Article 107)				
		PEL	0.05 mg/m3	California permissible exposure				
				limits for chemical contaminants				
			5400	(Title 8, Article 107)				
		see Section 5198						

Biological occupational exposure limits

Biological occupational exposure limits						
Component	CAS-No.	Parameters	Value	Biological specimen	Basis	
	-	Fluoride	2 mg/l	Urine	ACGIH - Biological Exposure Indices (BEI)	
	Remarks	Prior to shift (16 hours after exposure ceases)				

Aldrich - 236152 Page 4 of 9

Fluoride	3 mg/l	Urine	ACGIH - Biological Exposure Indices (BEI)
End of shift (As soon as possible after exposure ceases)			
Lead	200 μg/l	In blood	ACGIH - Biological Exposure Indices (BEI)
Not critical			

8.2 Exposure controls

Appropriate engineering controls

Avoid contact with skin, eyes and clothing. Wash hands before breaks and immediately after handling the product.

Personal protective equipment

Eye/face protection

Face shield and safety glasses Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU).

Skin protection

Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique (without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands.

Full contact

Material: Nitrile rubber

Minimum layer thickness: 0.11 mm Break through time: 480 min

Material tested:Dermatril® (KCL 740 / Aldrich Z677272, Size M)

Splash contact

Material: Nitrile rubber

Minimum layer thickness: 0.11 mm Break through time: 480 min

Material tested: Dermatril® (KCL 740 / Aldrich Z677272, Size M)

data source: KCL GmbH, D-36124 Eichenzell, phone +49 (0)6659 87300, e-mail sales@kcl.de, test method:

EN374

If used in solution, or mixed with other substances, and under conditions which differ from EN 374, contact the supplier of the CE approved gloves. This recommendation is advisory only and must be evaluated by an industrial hygienist and safety officer familiar with the specific situation of anticipated use by our customers. It should not be construed as offering an approval for any specific use scenario.

Body Protection

Complete suit protecting against chemicals, The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace.

Respiratory protection

Where risk assessment shows air-purifying respirators are appropriate use a full-face particle respirator type N100 (US) or type P3 (EN 143) respirator cartridges as a backup to engineering controls. If the respirator is the sole means of protection, use a full-face supplied air respirator. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU).

Control of environmental exposure

Prevent further leakage or spillage if safe to do so. Do not let product enter drains. Discharge into the environment must be avoided.

9. PHYSICAL AND CHEMICAL PROPERTIES

9.1 Information on basic physical and chemical properties

a) Appearance Form: powder Colour: white

b) Odourc) Odour Thresholddata availableNo data available

Aldrich - 236152 Page 5 of 9

d) pH No data available

e) Melting point/freezing

point

Melting point/range: 824 °C (1,515 °F) - lit.

f) Initial boiling point and

boiling range

explosive limits

No data available

g) Flash point Not applicable

No applicable

No data available

h) Evaporation rate No data availablei) Flammability (solid, gas) No data available

j) Upper/lower No data available flammability or

k) Vapour pressure No data availablel) Vapour density No data available

m) Relative density 8.445 g/mL at 25 °C (77 °F)

n) Water solubility No data available
 o) Partition coefficient: n- octanol/water

No data available

p) Auto-ignition temperature

No data available

q) Decomposition temperature

No data available

r) Viscosity No data available
s) Explosive properties No data available
t) Oxidizing properties No data available

9.2 Other safety information

No data available

10. STABILITY AND REACTIVITY

10.1 Reactivity

No data available

10.2 Chemical stability

Stable under recommended storage conditions.

10.3 Possibility of hazardous reactions

No data available

10.4 Conditions to avoid

Reacts dangerously with glass.

10.5 Incompatible materials

glass

10.6 Hazardous decomposition products

Hazardous decomposition products formed under fire conditions. - Hydrogen fluoride, Lead oxides Other decomposition products - No data available

In the event of fire: see section 5

11. TOXICOLOGICAL INFORMATION

11.1 Information on toxicological effects

Acute toxicity

LD50 Oral - Rat - 3,031 mg/kg

Remarks: Behavioral:Muscle weakness. Lungs, Thorax, or Respiration:Respiratory stimulation. Skin and Appendages: Other: Hair.

Aldrich - 236152 Page 6 of 9

Inhalation: No data available

No data available

Skin corrosion/irritation

No data available

Serious eye damage/eye irritation

No data available

Respiratory or skin sensitisation

No data available

Germ cell mutagenicity

No data available

Carcinogenicity

IARC: 3 - Group 3: Not classifiable as to its carcinogenicity to humans (Lead difluoride)

2A - Group 2A: Probably carcinogenic to humans (Lead difluoride)

NTP: RAHC - Reasonably anticipated to be a human carcinogenThe reference note has been

added by TD based on the background information of the NTP. (Lead difluoride)

OSHA: OSHA specifically regulated carcinogen (Lead difluoride)

Reproductive toxicity

No data available

Reproductive toxicity - Rat - Inhalation

Paternal Effects: Spermatogenesis (including genetic material, sperm morphology,motility, and count). Effects on Embryo or Fetus: Fetotoxicity (except death, e.g., stunted fetus).

Reproductive toxicity - Rat - Inhalation

Effects on Fertility: Post-implantation mortality (e.g., dead and/or resorbed implants per total number of implants). Effects on Embryo or Fetus: Fetal death.

Known human reproductive toxicant

No data available

Specific target organ toxicity - single exposure

No data available

Specific target organ toxicity - repeated exposure

May cause damage to organs through prolonged or repeated exposure.

Aspiration hazard

No data available

Additional Information

RTECS: OG1225000

Fluoride ion can reduce serum calcium levels possibly causing fatal hypocalcemia.

Lead salts have been reported to cross the placenta and to induce embryo- and feto- mortality. They also have teratogenic effect in some animal species. No teratogenic effects have been reported with exposure to organometallic lead compounds. Adverse effects of lead on human reproduction, embryonic and fetal development, and postnatal (e.g., mental) development have been reported. Excessive exposure can affect blood, nervous, and digestive systems. The synthesis of hemoglobin is inhibited and results in anemia. If left untreated, neuromuscular dysfunction, possible paralysis, and encephalopathy can result. Additional symptoms of overexposure include: joint and muscle pain, weakness of the extensor muscles (frequently the hand and wrist), headache, dizziness, abdominal pain, diarrhea, constipation, nausea, vomiting, blue line on the gums, insomnia, and metallic taste. High body levels produce increased cerebrospinal pressure, brain damage, and stupor leading to coma and often death., Kidney injury may occur.

Stomach - Irregularities - Based on Human Evidence Stomach - Irregularities - Based on Human Evidence

Aldrich - 236152 Page 7 of 9

12. ECOLOGICAL INFORMATION

12.1 Toxicity

No data available

12.2 Persistence and degradability

Biodegradability Result: - Not readily biodegradable.

12.3 Bioaccumulative potential

No data available

12.4 Mobility in soil

No data available

12.5 Results of PBT and vPvB assessment

PBT/vPvB assessment not available as chemical safety assessment not required/not conducted

12.6 Other adverse effects

An environmental hazard cannot be excluded in the event of unprofessional handling or disposal. Very toxic to aquatic life with long lasting effects.

13. DISPOSAL CONSIDERATIONS

13.1 Waste treatment methods

Product

Offer surplus and non-recyclable solutions to a licensed disposal company. Contact a licensed professional waste disposal service to dispose of this material. Dissolve or mix the material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber.

Contaminated packaging

Dispose of as unused product.

14. TRANSPORT INFORMATION

DOT (US)

UN number: 2291 Class: 6.1 Packing group: III Proper shipping name: Lead compounds, soluble, n.o.s. (Lead difluoride)

Reportable Quantity (RQ): 10 lbs Poison Inhalation Hazard: No

IMDG

IATA

UN number: 2291 Class: 6.1 Packing group: III Proper shipping name: Lead compound, soluble, n.o.s. (Lead difluoride)

15. REGULATORY INFORMATION

SARA 302 Components

No chemicals in this material are subject to the reporting requirements of SARA Title III, Section 302.

SARA 313 Components

This material does not contain any chemical components with known CAS numbers that exceed the threshold (De Minimis) reporting levels established by SARA Title III, Section 313.

SARA 311/312 Hazards

Acute Health Hazard, Chronic Health Hazard

Massachusetts Right To Know Components

Lead difluoride CAS-No. Revision Date 7783-46-2 1993-02-16

Pennsylvania Right To Know Components

Lead difluoride CAS-No. Revision Date 1993-02-16

Aldrich - 236152 Page 8 of 9

Lead difluoride CAS-No. Revision Date 7783-46-2 1993-02-16

New Jersey Right To Know Components

Lead difluoride CAS-No. Revision Date 7783-46-2 1993-02-16

California Prop. 65 Components

WARNING! This product contains a chemical known to the State of California to cause cancer. CAS-No. Revision Date 2007-09-28

Lead difluoride

16. OTHER INFORMATION

Full text of H-Statements referred to under sections 2 and 3.

Acute Tox. Acute toxicity

Aquatic Acute Acute aquatic toxicity
Aquatic Chronic Chronic aquatic toxicity

Carc. Carcinogenicity
H300 Fatal if swallowed.

H300 + H330 Fatal if swallowed or if inhaled.

H330 Fatal if inhaled. H350 May cause cancer.

H360 May damage fertility or the unborn child.

H373 May cause damage to organs through prolonged or repeated exposure.

HMIS Rating

Health hazard: 4
Chronic Health Hazard: *
Flammability: 0
Physical Hazard 0

NFPA Rating

Health hazard: 4
Fire Hazard: 0
Reactivity Hazard: 0

Further information

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Preparation Information

Sigma-Aldrich Corporation Product Safety – Americas Region 1-800-521-8956

Version: 4.14 Revision Date: 04/13/2018 Print Date: 10/04/2018

Aldrich - 236152 Page 9 of 9